



Enzymatic C-Methylation

Dual Role of S-Adenosylmethionine (SAM⁺) in the Methylation of sp²-Hybridized Electrophilic Carbons

Wolfgang Buckel and Rudolf K. Thauer*

active sites \cdot radicals \cdot reaction mechanisms \cdot transferases

M any biomolecules contain C-bound methyl groups that are formed from C_1 units; examples include thymidine and ubiquinone. In these compounds the methyl groups are introduced through S_N2 methylene- and methyl-group-transfer reactions from methylenetetrahydrofolate (methylene- H_4F)^[1] and S-adenosylmethionine (SAM⁺),^[2] respectively. Other cases of C-methyl groups generated by C_1 transfer to nucleophilic carbon atoms are the SAM⁺-derived C-methyl groups in 5-methylcytosine^[3] and menaquinone (methylnaphthoquinone)^[2] as well as the two methyl groups in precorrin-2, an intermediate in the biosyntheses of vitamin B_{12} , siroheme, heme D, and factor 430 (F_{430}) of the methanogenic Archaea.^[4]

There are, however, many compounds that are C-methylated with SAM+ without a nucleophilic carbon for methylgroup acceptance. Such a compound is adenosine 2503 in the 23S rRNA of bacterial ribosomes, whose posttranscriptional methylation at C-2 has a housekeeping function important in translational fidelity and whose methylation at C-8 confers resistance of bacteria to several classes of antibiotics that target the large subunit of the ribosome. C-2 and C-8 of adenosine are electrophilic rather than nucleophilic sp²-hybridized carbon atoms with poor C–H acidity, but despite this they are methylated by the SAM+-dependent methyltransferases RlmN from *Escherichia coli* and Cfr from *Staphylococcus aureus*, respectively.^[5]

The first clue to the mechanism came from the finding that the methyltransferases catalyzing C-2 and C-8 methylations belong to the radical SAM⁺ family of iron–sulfur proteins^[5a] and by that differ from those enzymes involved in S_N2 methylation reactions. The radical SAM⁺ enzymes generally catalyze reactions triggered by the 5'-deoxyadenolyl radical (5'-dA') that is generated by one-electron reduction of SAM⁺ which is bound to a [4Fe-4S] cluster.^[6] The second clue was the observation that the C-2 and C-8 methyl groups are derived from the methyl group of SAM⁺ and that in the methyl-transfer reaction S-adenosylhomocysteine (SAH) and 5'-deoxyadenosine (dA-H) plus methionine are formed.^[5a] And finally, the third clue was the observation that methylgroup transfer to C-2 and C-8 from methyl S-adenosyl[methyl-²H₃]methionine takes place with loss of one deuterium

atom, which is replaced by the hydrogen of the respective amidine carbon of the adenosine (Figure 1). [5b,7] In contrast, the SAM $^+$ -dependent methyl transferases, which work by an S $_{\rm N}2$ mechanism, catalyze the methyl-group transfer without exchange of any of the three hydrogen atoms.

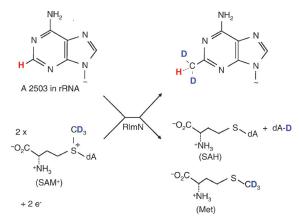


Figure 1. Formation of C-2-methyladenosine in 23S rRNA at position 2503 from SAM⁺ and 23S rRNA as catalyzed by the radical SAM⁺ methyltransferase RlmN from E. coli. Formation of C-8-methyl-A2503 is catalyzed analogously by the radical SAM⁺ methyltransferase Cfr from Staphylococcus aureus. RlmN and Cfr are phylogenetically closely related (33% sequence identity). dA-D = 5'-deoxyadenosine containing one deuterium in the methyl group.

These and other experimental findings indicate that methylation of C-2 of adenosine proceeds in two steps consuming two molecules of SAM+ per methyl group transferred. In step 1, one of the two essential cysteine thiol groups in the active site of the methyltransferase (HS-MT-SH) is methylated by SAM⁺ in an S_N2 reaction yielding HS-MT-S-CH₃ and SAH (reaction 1).^[7] In step 2, a second SAM⁺ molecule which is bound to the [4Fe-4S] cluster, is reduced by one electron to give methionine (Met) and 5'-dA', which subsequently abstracts a hydrogen from the protein-bound methyl group to yield 5'-dA-H and a neutral, carbon-centered methylene radical. Attack of this radical at the sp²-hybridized C-2 of adenosine 2503 (A 2503) in rRNA, removal of an electron, and deprotonation of the respective amidine carbon yields a thioether that is reduced to C-2 methyl-A2503 by an adjacent cysteine (reaction 2).^[7] The catalytic cycle is closed by reduction of the disulfide bond (reaction 3) lastly provid-

[*] Prof. Dr. W. Buckel, Prof. Dr. R. K. Thauer Max-Planck-Institut für terrestrische Mikrobiologie Karl-von-Frisch-Strasse 10, 35043 Marburg (Germany) E-mail: thauer@mpi-marburg.mpg.de



ing the two electrons required for the reduction of SAM⁺ to Met and 5'-dA-H (reaction 4 = reactions 1 + 2 + 3).^[7] The methylation at C-8 proceeds by the same mechanism.

$$SAM^{+} + HS-MT-SH \rightarrow SAH + HS-MT-S-CH_{3} + H^{+}$$
 (1)

$$\begin{split} \text{SAM}^+ + \text{HS-MT-S-CH}_3 + \text{A2503} &\rightarrow \text{dA-H} + \text{Met} \\ + (\text{S-MT-S})_{\text{disulfide}} + \text{CH}_3\text{-A2503} + \text{H}^+ \end{split} \tag{2}$$

$$(S-MT-S)_{disulfide} + 2e^{-} + 2H^{+} = HS-MT-SH$$
(3)

$$2 \text{ SAM}^+ + 2 e^- + A2503 = \text{SAH} + 5' - \text{AH} + \text{Met} + \text{CH}_3 - \text{A2503}$$
 (4)

It has not yet been possible to show that the methyl-transferase can catalyze reaction 3. In the experiments the Cys355-methylated enzyme was incubated with SAM⁺, an rRNA fragment, and an electron donor, yielding methyladenosine and the disulfide (S-MT-S) as products, which corresponds to only half of the catalytic cycle.

The crystal structures of RlmN and RlmN with SAM⁺ reveal that a single molecule of SAM⁺ coordinates the [4Fe-4S] cluster. Residue Cys355 is S-methylated and located proximal to the SAM⁺ methyl group, suggesting that SAM⁺ involved in the initial methyl transfer (reaction 1) binds to the same site.^[8] Thus, RlmN takes advantage of the two faces of SAM^{+[9]} in one reaction with structural economy.

C-2 and C-8 methylations of adenosine in rRNA are not the only examples of methylations of sp²-hybridized electrophilic carbons with SAM⁺. Others include the methylation of C-2 of tryptophan in thiostrepton^[10] and the synthesis of cyclopropane fatty acids from mono-unsaturated fatty acids and SAM+.[11] However, there is evidence that these two methylations do not follow the mechanism involved in the formation of C-2- and C-8-methyladenosine in rRNA. In the case of thiostrepton, which may also employ radical chemistry, it was shown that the methyl group in 2-methyltryptophan is incorporated from SAM+ with retention of configuration at the methyl carbon.^[10] This excludes mechanisms proceeding either by an S_N2 methyl-group transfer with inversion of configuration or via a methylene radical with hydrogen exchange. In the case of the cyclopropane fatty acids, the analysis of the responsible genes revealed the absence of the characteristic [4Fe-4S] binding motif of radical SAM+ enzymes (CX_3CX_2C , whereby C = cysteine and X = any amino acid).[11a]

There are also sp³-hybridized electrophilic carbons that can be methylated, [12] for example C-8² and C-12¹ in bacteriochlorophyll c (Figure 2), whose methylations involve radical SAM⁺ enzymes. [13] Other examples are the synthesis of the C-alkyl groups in pactamycin, [14] thienamycin, sitosterol, stigmasterol, and related 24-ethyl steroids, the synthesis of C-5-methylarginine and C-2-methylglutamine in methylcoenzyme M reductase (MCR) from methanogenic Archaea, [15] the synthesis of the two C-methyl groups in tetrahydromethanopterin, [16] and the synthesis of the C-methyl group in fosfomycin. [17] In the case of the two amino acids in MCR [15] and of tetrahydromethanopterin [16] it was shown that the methyl groups are introduced from [methyl-²H₃]methionine without D/H exchange by enzymes that still remain to be identified. [18]

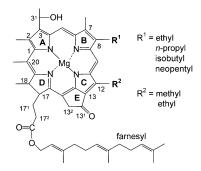


Figure 2. Structure of bacteriochlorophyll c from *Chlorobaculum tepidum*. The side chains R^1 and R^2 arise from successive methylations at the unreactive C-8² and C-12¹ methyl groups with SAM⁺ as the donor. [13]

A mechanistic proposal for the methylation of the sp³-hybridized carbon atom has been put forward for the methylation reaction in fosfomycin. [12b] It is based on the finding that the gene fom3 involved encodes for a protein with two domains. The conserved N-terminal domain is annotated as B_{12} -cofactor-binding, whereas the C-terminal domain shows a high degree of sequence similarity to the enzyme of the radical SAM⁺ family, including the diagnostic [4Fe-4S] cluster binding motif. The hypothesis is that a 5'-deoxyade-nosyl radical generates a carbon-centered substrate radical, which can subsequently react with the methyl cobalamin to form the methylated product. This would be the first example of a radical methyl transfer from methylcobalamin (methyl- B_{12}) as the donor. Thus, much remains much to be discovered in this area.

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